

THE CATHOLIC UNIVERSITY OF EASTERN AFRICA

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MAIN EXAMINATION

JANUARY - APRIL 2015 TRIMESTER

FACULTY OF SCIENCE

DEPARTMENT OF CHEMISTRY

SCHOOL FOCUSED PROGRAMME

SPECIAL EXAMINATION

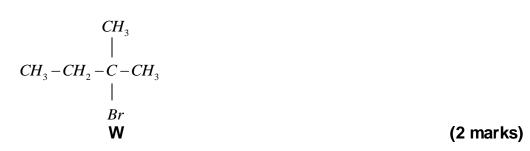
CHEM 204: REACTION MECHANISM

Date: APRIL 2015 Duration: 2 Hours			
INSTRUCTION	S: Answer Question ONE and ANY OTHER	TWO Questions	
Q1. a) i)	An organic compound (A) on analysis gave hydrogen 6.6% and the rest was oxygen. I) Calculate the empirical formula of compound (C = 12.0 H = 1.0 O = 16.0) II) Given that the molecular mass of compound or compound the possible isom	empound A. mpound A is 90.0 g.	
ii)	A certain alcohols first oxidized to compethen to $C_2H_4O_2(Y)$. I) Write the equation for the reaction. II) Name compound X and Y.	2 7	
iii)	Z.I) Write an equation for the reaction.II) Name compound Z	•	

- iv) Compound Y reacts with ethanol in the presence of an inorganic acid to form a sweet smelling compound. Write a mechanism for the reaction.
- b) Explain the following observations:
 - i) The general trend in the stability of carbonations is $3^{\circ} > 2^{\circ} > 1^{\circ}$ (1 mark)
 - ii) Benzene undergoes electrophilic substitution whereas alkenes undergo addition reaction. (2 marks)
- c) Define the following terms giving examples in each case:
 - i) Homolytic bond fission
 - ii) Electrophiles
 - iii) Nucleophiles
 - iv) Elimination

(3 marks)

d) Which of the following alkyl halides would you expect to undergo unimolecular substitution reaction. Explain your answer.



- e) Predict the necessary information (products or reagents to complete the following:
 - i) Benzene → Bromobenzene
 - ii) Propanone + hydrazine → ?

Write a plausible mechanism for each reaction. (7 marks)

Q2. Discuss **FIVE** major organic reactions given specific examples in each case. (20 marks)

- Q3. Compare and contrast the mechanism of substitution reactions in:
 - a) alkanes
 - b) alkylhalides

Give mechanisms in each case.

(20 marks)

Q4. Discuss the reduction and oxidation of organic compounds containing oxygen. Give specific examples and mechanisms in each case.

(20 marks)

Q5. Describe the reactions of benzene given the mechanism of each reaction described. (20 marks)

END