**\* THE CATHOLIC UNIVERSITY OF EASTERN AFRICA** 



# A. M. E. C. E. A

MAIN EXAMINATION

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### JANUARY – APRIL 2019 TRIMESTER

## FACULTY OF SCIENCE

#### DEPARTMENT OF NATURAL SCIENCE (CHEMISTRY)

#### **REGULAR PROGRAMME**

#### CHEM 310: ENVIRONMENTAL CHEMISTRY

# Date: APRIL 2019Duration: 2 HoursINSTRUCTIONS: Answer Question ONE and ANY OTHER TWO Questions

Q1.	i)	Define the following terms used in environmental chemistry a) Sink b) Pathway of a pollutant c) Speciation d) Eutrophication e) Bioamplification [8 marks]	[10 marks]
	ii) [8 <i>ma</i> iii)	Discuss the importance of the Atmosphere [rks] Distinguish among geosphere, lithosphere and crust of the e	[10 <i>marks</i> ] earth
[6mar	ks] [8 ma iv)	If the RfD (reference dose = maximum oral dose of toxic sub methylmercury is 0.1μg kg <sup>-1</sup> body weight/day, what mass of person safely eat each week if the average methyl mercury fish is 0.3 μg/g [8 marks]	fish can 60 kg
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Q2. a) The COD of a water sample is 25 mg of  $O_2$  per litre. What volume of 0.0010 mol L<sup>-1</sup> Na<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub> solution is required to titrate a 40 mL sample to end-point [8 marks]

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b) Explain measures that can be put in place to control water pollution c) 3 marks Q3. Discuss the role of CO<sub>2</sub> in the atmosphere and in the Hydrosphere a) 12 marks 8 marks What are the implications of increasing the concentration of CO<sub>2</sub> in these b) environments? 8 marks

8 marks

- Q4. Discuss the consequences of green house effects and the abatement of the green house effects 20 marks 8 marks
- Differentiate between classical smog from photo chemical smog Q5. a)

6 marks

- Using appropriate chemical equations to discuss the formation of b) photochemical smog 8 marks
- Explain the effects of photochemical smog and control measures used to c) reduce the photochemical smog. 6 marks

\*END\*

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Explain the major causes of water pollution