THE CATHOLIC UNIVERSITY OF EASTERN AFRICA

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MAIN EXAMINATION

MAY – JULY 2015 TRIMESTER

FACULTY OF SCIENCE

DEPARTMENT OF CHEMISTRY

CHEM 101: ORGANIC CHEMISTRY I

SCHOOL FOCUSED PROGRAMME

Date: JULY 2015 Duration: 2 Hours

INSTRUCTIONS: Answer Question ONE and ANY OTHER TWO Questions

Q1. a) Draw the structures of the following compounds.

i Trans-2-butene

ii Pentan-2-one

iii Propylethanoate

iv 3-iodocyclopentene

v 4-(I-methylethyl) - 5,5 - dimethylnonane (5 marks)

b) Give the IUPAC names of the following

i
$$CH_3$$
— CH_2 — CH — CH_3

iii CH₃COOCH₂CH₂CH₃

$$_{\mathrm{iv}}$$
 $\overset{\mathsf{OH}}{\overset{\mathsf{OH}}{\overset{\mathsf{CH}_{3}}{\overset{\mathsf{C}}{\overset{\mathsf{CH}_{3}}{\overset{\mathsf{C}}{\overset{\mathsf{CH}_{3}}{\overset{\mathsf{C}}{\overset{\mathsf{CH}_{3}}{\overset{\mathsf{C}}{\overset{\mathsf{CH}_{3}}{\overset{\mathsf{C}}{\overset{\mathsf{C}}{\overset{\mathsf{CH}_{3}}{\overset{\mathsf{C}}}{\overset{\mathsf{C}}{\overset{\mathsf{C}}{\overset{\mathsf{C}}{\overset{\mathsf{C}}{\overset{\mathsf{C}}{\overset{\mathsf{C}}{\overset{\mathsf{C}}{\overset{\mathsf{C}}}{\overset{\mathsf{C}}{\overset{\mathsf{C}}}{\overset{\mathsf{C}}{\overset{\mathsf{C}}}{\overset{\mathsf{C}}{\overset{\mathsf{C}}}{\overset{\mathsf{C}}{\overset{\mathsf{C}}}{\overset{\mathsf{C}}}{\overset{\mathsf{C}}{\overset{\mathsf{C}}}{\overset{\mathsf{C}}{\overset{\mathsf{C}}}{\overset{\mathsf{C}}{\overset{\mathsf{C}}}{\overset{\mathsf{C}}{\overset{\mathsf{C}}}{\overset{\mathsf{C}}}{\overset{\mathsf{C}}}{\overset{\mathsf{C}}}{\overset{\mathsf{C}}}{\overset{\mathsf{C}}}{\overset{\mathsf{C}}}{\overset{\mathsf{C}}}{\overset{\mathsf{C}}}{\overset{\mathsf{C}}}{\overset{\mathsf{C}}}{\overset{\mathsf{C}}}{\overset{\mathsf{C}}}}{\overset{\mathsf{C}}}{\overset{\mathsf{C}}}{\overset{\mathsf{C}}}{\overset{\mathsf{C}}}{\overset{\mathsf{C}}}}{\overset{\mathsf{C}}}}{\overset{\mathsf{C}}}{\overset{\mathsf{C}}}{\overset{\mathsf{C}}}}{\overset{\mathsf{C}}}}{\overset{\mathsf{C}}}}{\overset{\mathsf{C}}}}{\overset{\mathsf{C}}}{\overset{\mathsf{C}}}}{\overset{\mathsf{C}}}}{\overset{\mathsf{C}}}}{\overset{\mathsf{C}}}{\overset{\mathsf{C}}}}{\overset{C}}}{\overset{C}}}{\overset{C}}}{\overset{C}}}{\overset{C}}}{\overset{C}}}{\overset{C}}}{\overset{C}}{\overset{C}}}{\overset{C}}}{\overset{C}}}{\overset{C}}}{\overset{C}}}{\overset{C}}}{\overset{C}}{\overset{C}}}{\overset{C}}}{\overset{C}}}{\overset{C}}}{\overset{C}}}{\overset{C}}}{\overset{C}}}{\overset{C}}{\overset{C}}}{\overset{C}}}{\overset{C}}}{\overset{C}}}{\overset{C}}{\overset{C}}}{\overset{C}}}{\overset{C}}}{\overset{C}}}{\overset{C}}{\overset{C}}}{\overset{C}}}{\overset{C}}{\overset{C}}}{\overset{C}}}{\overset{C}}}{\overset{C}}{\overset{C}}{\overset{C}}}{\overset{C}}}{\overset{C}}}{\overset{C}}{\overset{C}}{\overset{C}}}{\overset{C}}}{\overset{C}}}{\overset{C}}}{\overset{C}}}{\overset{C}}{\overset{C}}}{\overset{C}}{\overset{C}}}{\overset{C}}}{\overset{C}}}{$

$$\begin{array}{ccc}
 & & CH_3 & C & \longrightarrow C & CH_3 \\
 & & & c & d
\end{array}$$

(5 marks)

c) Indicate the type of hybridization of each carbon atom in the following compound.

$$CH_2$$
 CH_3 CCH_3 CCH_3 CCH_3

(2 marks)

- d) Draw structures of molecules with the formular C₄H₈O that contain
 - i An alcol
 - ii An ether
 - iii A ketone
 - iv An aldehyde.

(4 marks)

e) Complete the following reactions by writing the missing structures A-F

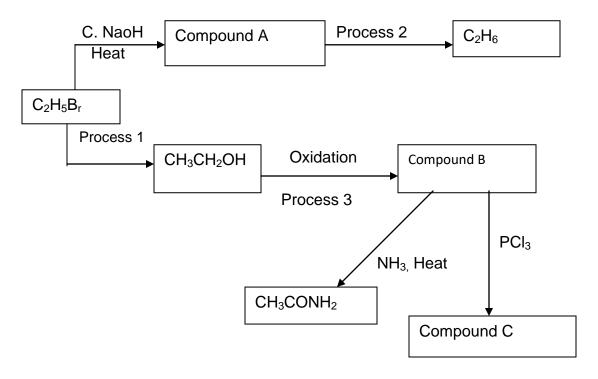
ii
$$CH_2CH_2 + HCI \longrightarrow ?B$$

IV
$$HC \equiv CH + H_2 \xrightarrow{Ni, 200^0 c}$$
 ? D

V $CH_3CHO + NH_2NH_2 \xrightarrow{}$? E

Vi $CH_3COOCH_2 + ? F \xrightarrow{}$ $CH_3COOCH_2CH_2CH_3$

- f) Show all the hydrogen bonds that will form in an aqueous solution of ethanoic acid. (2 marks)
- g) Study the flow chart below and answer the questions that follow



- i Draw structures of compound A, B, and C. (3 marks)
- ii Name processes 1, 2 and 3 (3 marks)
- Q2. Aldehydes and Ketones are carbonyl compound and have a general formular $C_n H_{2n}O$. Discuss the chemistry of carbonyl compounds under the following sub headings
 - i Structure and nomenclature.

(4 marks)

	ii	Preparation.	(3 marks)
	iii	Chemical reactions	(10 marks)
	iv	Uses.	(2 marks)
Q3.	Discuss the chemistry of alkyhalides and alcohols under the following subheadings;		
	i	Isomerism and classification.	(6 marks)
	ii 	Preparation	(4 marks)
	iii	Chemical reactions and test.	(10 marks)
Q4.	a)	Discuss the chemistry of alkynes under the following subheat Preparation	adings (3 marks)
		ii Chemical reactions that are similar to alkenes.	(8 marks)
		iii Chemical reactions that are different from alkenes.	(4 marks)
	b)	Describe the fractional distillation process of crude oil.	(5 marks)
Q5.	a)	Compare the chemistry of carboxylic acids to that of alcohols similarities and differences).	s. (Give (10 marks)
	b)	Discuss the chemistry of hybdridization in hydrocarbons (alk and alkynes)	anes, alkenes