



THE CATHOLIC UNIVERSITY OF EASTERN AFRICA

A. M. E. C. E. A

MAIN EXAMINATION

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AUGUST – DECEMBER 2018 TRIMESTER

FACULTY OF SCIENCE

DEPARTMENT OF NATURAL SCIENCE (CHEMISTRY)

PART TIME PROGRAMME

CHEM 310: ENVIRONMENTAL CHEMISTRY

Date: DECEMBER 2018

Duration: 2 Hours

INSTRUCTIONS: Answer Question ONE and ANY OTHER TWO Questions

- Q1. i). Define the following terms used in environmental chemistry
- a) Source
 - b) Pathway of a pollutant
 - c) Receptor
 - d) Eutrophication
 - e) Bioamplification
- ii) Discuss the importance of the Atmosphere [10 marks]
- iii) Distinguish between the troposphere and the Stratosphere [6 marks]
- iv) If the RfD (reference dose = maximum oral dose of toxic substance) for methylmercury is $0.1 \mu\text{g kg}^{-1}$ body weight/day, what mass of fish can 60 kg person safely eat each week if the average methyl mercury level in the fish is $0.3 \mu\text{g/g}$ [4marks]
- Q2. a). The COD of a water sample is 25 mg of O_2 per litre. What volume of $0.0010 \text{ mol L}^{-1} \text{ Na}_2\text{Cr}_2\text{O}_7$ solution is required to titrate a 40 mL sample to end-point [8 marks]
- b). Explain the major causes of water pollution [9 marks]
- c). Explain measures that can be put in place to control water pollution [3 marks]

- Q3. a) Differentiate between classical smog from photo chemical smog
[6 marks]
- b) Using appropriate chemical equations discuss the formation of photochemical smog
[8 marks]
- c). Explain the effects of photochemical smog and control measures used to reduce the photochemical smog.
[6 marks]
- Q4. a). Explain why Carbon monoxide gas is more dangerous than carbon dioxide gas
[5 marks]
- b). Briefly explain why temperature decreases with altitude in the troposphere, but increases with altitude in the stratosphere.
[6 marks]
- Q5. Discuss the consequences of green house effects and the abatement of the green house effect
[20 marks]

END