



THE CATHOLIC UNIVERSITY OF EASTERN AFRICA

A. M. E. C. E. A

MAIN EXAMINATION

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AUGUST – DECEMBER 2018 TRIMESTER

FACULTY OF SCIENCE

DEPARTMENT OF NATURAL SCIENCE (CHEMISTRY)

CHEM 202: NUCLEAR AND RADIATION CHEMISTRY

Date: DECEMBER 2018

Duration: 2 Hours

INSTRUCTIONS: Answer Question ONE and ANY OTHER TWO Questions

- Q1. a) i) Define the term radioactivity. **(2 marks)**
ii) Using relevant examples and a Nuclear equation explain the factors that influence the stability of a nucleus. **(10 marks)**
- b) i) Define atomic mass unit. **(2 marks)**
ii) Show how 1 amu is converted into energy in MeV. **(6 marks)**
- c) Using a relevant diagram describe the forces that bind the nucleus together. **(10 marks)**
- Q2. a) A sample of radioactive material contains 2×10^{10} atoms. The half-life of the material is 4-days. Calculate
i) The fraction remaining after 9 days. **(5 marks)**
ii) The activity of the sample after 9 days. **(5 marks)**
- b) i) What is neutron activation analysis? **(4 marks)**
ii) Explain the principles behind carbon-14 dating technique. **(6 marks)**
- Q3. a) i) What is radiochemistry? **(4 marks)**
ii) Describe how radioactivity is used in the following cases
- Agriculture
- Biology
- Medicine

- Water technology **(12 marks)**
- b) Define radioactive equilibrium. **(4 marks)**
- Q4. a) Show that the activity of a radioactive sample is given by $A = A_0 e^{-\lambda t}$ and define all the terms. **(10 marks)**
- b) Give two types of nuclear fuel and for each state how it is obtained. **(10 marks)**
- Q5. a) Discuss the factors that influence the biological effect of a particular source of radiation. **(8 marks)**
- b) Explain the mechanisms of Fe^{++} dosimeter. **(8 marks)**
- c) List 4 types of radioactive decay series and for each state the initial and final product. **(4 marks)**

END